ARONOVICH, V. M.; FRAVDYUK, N. F.

"Behavior of fuel under irradiation."

report submitted for 3rd Intl Conf, Peaceful Uses of Atomic Energy, Geneva,
31 Aug-9 Sep 64.

PRAVDYUK, N. F.; KONOBEYEVSKIY, S. T.; ORLOV, M. L.

"Effect of some factors on hydrogenization and properties of zirconium alloys used for Jackets of heat-producing elements in water cooled power reactors."

report submitted for 3rd Intl Conf, Peaceful Uses of Atomic Energy, Geneva, 31 Aug-9 Sep 64.

	L 8571-66 EPF(n)-2/EWA(h)/EWP(z)/EWP(b)/T/EWT(m)/EWA(d)/EWP(w)/EWP(t) IJP(c) ACC NR: AT5023782 GG/WW/JD SOURCE CODE. UR/20000/Co/colors/	
	ACC NRI AT5023782 GG/WW/JD SOURCE CODE: UR/0000/63/000/000/000/000/	
	1 200000 000000000000000000000000000000	
	AUTHOR: Pravdyuk, N. F.; Amayev, A. D.; Platonov, P. A.; Kuznetsov, V. N.;	
į	ORG: none 44, 55 44, 55 44, 55 44, 55 44, 55	
	ORG: none	
	10	
	TITLE: Effect of neutron irradiation of the properties of structural materials	
	SOURCE: 1945 Sovieghoberies	
	SOURCE and Soveshchaniye po probleme Deystviye yadernykh izlucheniy na materialy. Mos-	
7	cow, 1960. Deystviye yadernykh izlucheniy na materialy. Mostion on materials); doklady soveshchaniya. Moscow, Izd-vo AN SSSR, 1962, 34-57	
	74 13550, 124-10 All SSR, 1962, 34-57	
1	TOPIC TAGS: neutron irradiation, structural material, low carbon steel, low alloy steel, austenitic steel, steel property, zirconium allow allow	
-	steel, austenitic steel, steel property, zirconium alloy, alloy property, radiation	
	ABSTRACT: The effect of Arradiation of the mechanical properties of low-carbon steels, low-alloy steels, austenitic steels, and zirconium 201-	
1	steels, low-alloy steels, austenitic steels, and zirconium alloys has been investigated at the Institute of Atomic Energy im I V	
	bility as structured their suit-	
l	bility as structural materials for use in reactors. Irradiation of low-carbon steel with a flux of 10 ¹⁹ or 10 ²⁰ neutron/cm ² at 160—2000 increased the steel yield strength and tensile strength, but substantially decreased ductility. For example, the steel with the steel of the steel with the steel of the steel with the steel of the stee	
	and tensile strength, but substantially as 150-2000 increased the steel yield strength	,
	gaulon of low-carbon stool deep or the stool deep of the stool example, the stool of the stool deep of	
	Certain conditions of irradiating low-carbon ferrite or ferritic-pearlitic steels	
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change their properties to such an extent that their utilization in reactors involves a risk. Toughness and NDT temperature, not strength, determine the fitness of materials for use in reactor vessels. Irraditation of steels at temperatures under 250C with a 1018 neutron/cm2 flux causes some changes in their mechanical properties; a 1020 neutron/cm2 flux induces the maximum change (this is especially pronounced in stainless austenitic steels). Irradiation at temperatures above 400C has virtually no effect on the mechanical properties of structural materials. Stainless austenitic steels and nickel-chromium-iron alloys irradiated at 100C maintain satisfactory ductility (elongation of at least 20%). Austenitic steels and zirconium and its alloys, cold worked prior to irradiation, combine strength with moderate ductility (elongation of at least 10%). Low-carbon steel, low-alloy steels, and other materials, with a relatively high content of boron after irradiation, become brittle; their elongation after irradiation with 1020 neutron/cm2 is low. However, under conditions of low irradiation, the utilization of these low-carbon and low-alloy steels at low temperatures is admissable. In making thickwall reactor vessels from these steels, the NTD temperature is the main factor for determining the acceptable irradiation dose. Orig. art. has: 19 figures and 3 tables.

SUB CODE: 11, 18/ SUBM DATE: 18Aug62/ ORIG REF: 005/ OTH REF: 001

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9235-66 EAT(m)/EWP(w)/EPF(n)-2/EWA(d)/EWP(v)/T/EWP(t)/EWP(k)/EWP(z)/EWP(b)/EVA(h)/-MJW/JD/HM/GG/ GS UR/0000/62/000/000/0058/0067 ACC NRi AT5023783 EWA(c) AUTHOR: Amayev, A. D.; Yefimov, A. V.; Platonov, P. A.; Pravdyuk, N. F.; Razov, I.A.; Khlebnikov, A. M. 71,5; Khlebnikov, A. M. 11.5 44 55 TITLE: Effect of neutron irradiation on the mechanical properties of heatresistant ferritic-pearlitic steels and on their welded joints < SOURCE: Soveshchaniye po probleme Deystviye yadernykh izlucheniy na materialy. Moscow, 1960. Deystviye yadernykh izlucheniy na materialy (The effect of nuclear radiation on materials doklady soveschaniya. Moscow, Izd-vo AN SSSR, 1962, 58-67 TOPIC TAGS: ferritic pearlitic steel, neutron irradiation, steel irradiation, steel property, weld property/25Kh2MFA steel, 12Kh2MFA steel ABSTRACT: The effect of neutron irradiation on the mechanical properties of ferritic-pearlitic steels and their welded joints has been investigated. Specimens ferritic-pearlitic steels and their weiged joints has been investigated. The steels of annealed and tempered 25Kh2MFAl and 12Kh2MFA chromium-molybdenum-vanadium steels of annealed and tempered 25Kh2MFAl and 12Kh2MFA chromium-molybdenum-vanadium steels. of annealed and tempered 25Kh2MFAl and 12Kh2MFA chromium-motypuenum-variation self-44 16, with 0.2% and 0.1% C, respectively, were irradiated at 80-305C with integrated 44 16, with 0.2% and 0.1% C, respectively, were irradiated at 80-305C with energy > 1). with 0.2% and 0.1% 0, respectively, were illiadiated at 00 3050 with energy > 1). neutron fluxes of 2.8 x 10^{17} —7.2 x 10^{19} n/cm² (35% of neutrons with energy > 1). Mechanical tests of both steels and of 12Kh2MFA steel welds showed that neutron irradiation increases strength and decreases ductility and notch toughness; but not as much as in 25KhNM steel or 20 steel irradiated under the same conditions. This shows that metal strengthened by means of alloying or heat treatment, plastic Card 1/2

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deformation) or dispersion hardening is less sensitive to irradiation. The mechanical properties of 12Kh2MFA steel welds obtained by manual or automatic welding
with 12KhMF or 12KhM electrode wires undergo practically the same change as the
base metal when irradiated with a 7·10¹⁹ n/cm² flux. However, this change is
slightly more pronounced in welds obtained with 12KhM wire, owing to its lower
content of alloying elements. Increasing the temperature of irradiation decreased
the radiation damage in all tested steels and diminished the degree of change in
mechanical properties, because the damage is partially or completely eliminated
by annealing. The highest temperatures at which no change of mechanical properties
of ferritic-pearlitic steels and their welds occurs under effect of irradiation
with neutron flux of the indicated intensity are 350—400C. At an irradiation
temperature of 100C, none of the tested steels attains the highest values of strength,
art. has: 9 figures and 4 tables.

[ND]

SUB CODE: 11, 20/ SUBM DATE: 18Aug62/ ORIG REF: 003

Card 2/2

L 10800-66 EVT(m)/EVP(w)/EPF(n)-2/EWA(d)/T/EWP(t)/EWP(z)/EWP(b)/EWA(h) LJP(c)
ACC NR: AT5023784 MJW/JD/JG/GG/GS SOURCE CODE: UR/0000/62/000/000/0068/0973 14.55 AUTHOR: Yefimov, A. V.; Kozhevnikov, O. A.; Nikolayev, V. A.; Pravdyuk Razov, I. A.; Khlebrikov, A. M. ORG: none TITLE: Effect of neutron irradiation on the mechanical properties of stainless austenitic steels of various strength 44,55 SOURCE: Soveshchaniye po probleme Devstviye vadernykh izlucheniy na materialy. Moscow, 1960 Deystviye yadernykh izlucheniy na materialy (The effect of nuclear radiation on materials); doklady soveshchaniya. Moscow, Izd-vo AN SSSR, 1962. TOPIC TAGS: austenitic steel, austenitic alloy steel, neutron irradiation, steel irradiation, steel property ABSTRACT: The effect of neutron irradiation on the mechanical properties of stainless austenitic steels has been investigated. 1Kh18N9T steel austenitized at 1000C or austenitized at this temperature and cold rolled with 25% elongation, and austenitic, dispersion-hardenable, chromium-nickel steel of the 18-22 type, alloyed with tungsten and titanium were irradiated with integrated fluxes of 7.4 x 10²⁰ or 2 x 10²⁰ n/cm² with energy > 1MeV at 100C, 300C, or 500C. Tests showed that irradiation of as-austenitized 1Kh18N9T steel at 100C with 7.4 x 10¹⁹ n/cm² increases the yield and tensile strengths by 101% and 24%, respectively, and decreases the elongation and

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notch toughness by 39% and 20%. The same irradiation increases the yield and tensile strengths of austenitized and cold-rolled 1Kh18N9T steel only by 27% and 21%, and decreases its elongation and notch toughness by 38% and 42%. Increasing the irradiation intensity from 7.4 x 10¹⁹ to 2.10²⁰ n/cm² has no effect on the properties of this steel. Increasing the temperature of irradiation with 7.4 x 10¹⁹ n/cm² from 100 to 300 to 500C decreases the yield strength of austenitized and cold-rolled steel by 11% and 30% below that of steel irradiated at 100C. The tensile strength drops in this case by 4 and 17%, but the elongation increases by 44 and 148%. The mechanical properties of stainless chromium-nickel steel alloyed with tungsten and titanium and austenitized and aged at 710C for 10 hours, do not change much under the effect of fast-neutron irradiation at 2 10²⁰ n/cm², except for the yield strength, which increases by 30%. Orig. art. has: 4 figures and 2 tables. [ND]

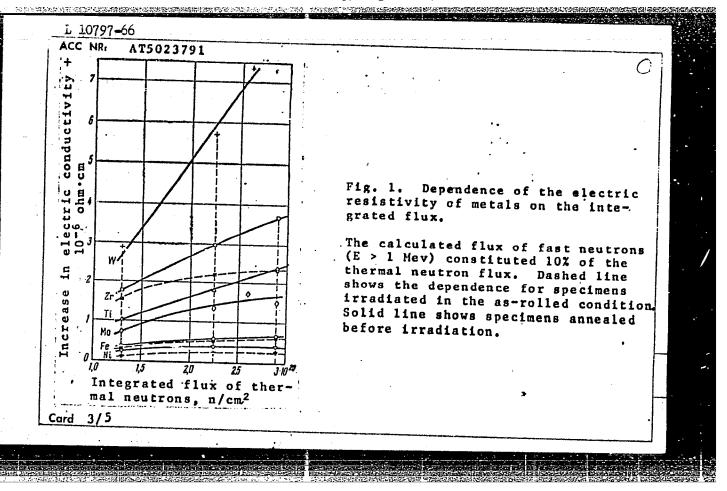
SUB CODE: /3, 20 SUBM DATE: 18Aug62/ ORIG REF: 003/ OTH REF: 008

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	L 10797-66 EWT(1)/EWT(m)/EPF(n)-2/T/EWP(t)/EWP(z)/EWP(b)/EWA(h)/EWA(c) IJP(c) ACC NR: AT5023791 SOURCE CODE: UR/2000/62/2000	
	JD/WW/HW/JG/GG/GS SOURCE CODE: UR/0000/62/000/000/0136/0152 AUTHOR: Ivanov, A. N.; Pravdyuk, N. F.	
	ORG: none	
	TITLE: Effect of	
	TITLE: Effect of neutron irradiation on the electric resistivity of	
	SOURCE: Soveshchaniye po probleme devstvive vadernykh izlucheniy na	
+	1446 CIICC OF DUCTOR WAS ALLE	- 1
	Moscow, Izd-vo AN SSSR, 1962, 136-152	
	TOPIC TAGS: iron, nickel, titanium, iodida zirconium, molybdenum, tungsten, neutron irradiation, metal electric	
	irradiation effect	
	ABSTRACT: The Atomic Energy Institute im. I. V. Kurchatov has investigated the effect of neutron irradiation at 40-500 on the electrical	
\downarrow	gated the effect of neutron irradiation at 40—50C on the electric is nium, molybdenum, and tungsten. The metals in the as-rolled and	
+	annealed and tungsten. The metals in the as-rolled	
	annealed conditions were irradiated with a flux of 2.0-2.5 x 1013 n/	
	n/cm ² ·sec (thermal) and about 2.0—2.5 x 10 ¹² n/cm ² ·sec fast neutrons	
	h method developed by the authors	
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•	Table 1.	Relative change i	n the e	leafyd a y	ont net ed			G 2
	metals de	pending on the int	egrated	flux.	COTOCTAT	Ly OF Variou	31	
•	ź.		Elec- trical		;		- [
	5.		resis- tivity prior				;	
	Materials	Preliminary treatment	to ir-	Change in (%) at a of 10 ²⁰		resistivity ited flux ermal)*	1	
		1	opni • cm	1,3	2,25	2,9		
	Iron {	Annealing at 700C, 1 hr Reduction up to 94%		2,6±0,3 2,1±0,2	4,0±0,1	4,8±0,1	 ,	
į	Nickel {	Annealing at 700C, 1 hr Reduction up to 92.5%	9,12 9,41	2,9*** 1,1±0.2	3,7±0,1 4,6±0,1 2,7±0,1	4,4±0,5** 4,9±0,1 3,1±0,1	:	
		Armealing at 1000 C, 1 hr Reduction up to 95.5%	51.63	3,6±0,6	6,1±0,3	7,6±0,1	•	
	. II Canium	Annealing at 1000C, 1hr Annealing at 1000C, 1hr	60.00	1,7±0,4 12,6±1,7**	4,3±0,3 3,0±0,1 22,3±0,45	.4,6±0,2 4,0±0,1	i i	
	Tungsten	Annealing at 1000 C, 1 hr	6,10	47.3±10.3	94+1 4**	28,9±0,4****	·	-
	non Sinnia	eutron flux consti	tuted a	bout 10%	of therma	1 neutrons) :	
ard 2/5	-pa	ens were irradiated 1020 n/cm2 therma	7 of 40.	50C wit	h an inte	grated flux		



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for the remote measurement of the electrical resistivity of metals 0 during irradiation in the reactor was used in the experiments. Results of the measurements are shown in Fig. 1 and Table 1. To determine the nature of the irradiation defects, the kinetics of the change in the resistivity of the metals irradiated with various integrated fluxes was investigated by means of isochronal annealing. It was found that the removal of the irradiation-induced increase in the resistivity of titanium, zirconium, and iron irradiated in the annealed conditions proceeded in a single stage and was complete at 210-290, 300-400, and 350C for T1, Zr, and Fe, respectively. This seems to indicate annealing not of elementary, but of more complex defects. In rolled irradiated and unirradiated iron, the removal of the resistivity increment proceeded in two stages: the first at 100-250C, associated with the release of simple defects from traps, and the second at 250-550C, associated with the rearrangement of dislocations. The decrease of the irradiation-induced increase in resistivity in irradiated molybdenum and tungsten also proceeded in two stages: at 100-250 and above 450C in molybdenum and at 100-375 and 375-1000C in tungsten. The low-temperature stage of the decrease appears to be associated with The way to be the west of the same. they and the life elementative purishwith the attributed of

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SUB	CODE:	13,20	SUBM	DATE:	18Aug62/	ORIG	REF:	000/	отн	REF:	015	
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EPF(n)=2/EWT(1)/EWT(m)/T/EWP(b)/EWA(h)/EWP(w)/EWP(t)ACC NR: AT5023792 SOURCE CODE: UR/0000/62/000/000/0153/0159 AUTHOR: Pravdyuk, N. F.; Platonov, P. A जं स ORG: none Investigation of the rupture strength of copper after irradiation TITLE: SOURCE: N Soveshchaniye po probleme Deystviye yadernykh izlucheniy na materialy. Moscow, 1960. Deystviye yadernykh izlucheniy na materialy (The effect of nuclear radiation on materials); doklady soveshchaniya. Moscow, Izd-vo AN SSSR, 1962, TOPIC TAGS: copper, fast neutron irradiation, neutron irradiated copper, copper rupture strength, irradiation effect 21, 90, 5 ABSTRACT: Commercial copper was irradiated with a fast neutron flux of about 10^{20} n/cm² at 80C. Then both unirradiated and irradiated copper were subjected to stress-rupture tests at 200-300C at stresses varying from 4 to 19 kg/mm². The rupture life in irradiated and unirradiated copper decreased linearly as stress increased. Under identical stresses and temperatures, the rupture life of irradiated copper was many times longer than that of unirradiated. Investigation of the rupture life-temperature dependence showed that the failure of both unirradiated and irradiated copper occurred with an identical value of the activation energy, which decreased with increasing stress. Analysis of the data obtained showed that the failure of copper in both states occurred with the same mechanism. Presumably, the

ACC NR. AT5023792 rupture life would increase with irradiation at low temperatures, when diffusion processes pley an insignificant role and the rupture life is determined mainly by the intensity of the formation of vacancies resulting from plastic deformation during irradiation. It can then be expected that at temperatures equal to or higher than the melting temperature and at lower stresses (consequently, at a longer rupture life) irradiation will shorten the rupture life. This circumstance, however, requires experimental verification. Orig. art. has: 5 figures and 3 tables. [M5] SUB CODE: 11,18/ SUBM DATE: 18Aug62/ ORIG REF: 003/ OTH REF: 001

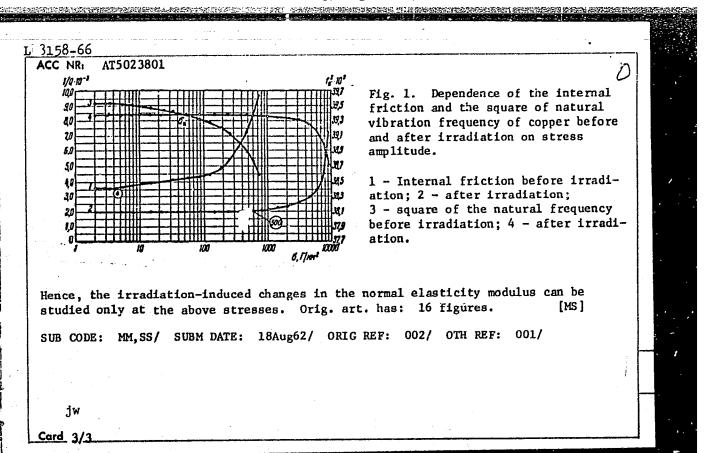
L 8158-66 EPF(n)-2/EWT(d)/EWT(m)/EWP(z)/EWP(b)/T/EWA(d)/EWP(w)/EW1(t)-1 ACC NR: AT5023801 EM/GG/MJW/JD/HW/GS SOURCE CODE: UR/0000/62/000/000/0219/0 AUTHOR: Konobeyevskiy, S. T. (Corresponding member AN SSSR); Pravdyuk, Pokrovskiy, Yu. I.; Vikhrov, 44.55 TITLE: The effect of neutron irradiation on the internal friction of metals SOURCE: Soveshchaniye po probleme Deystviye yadernykh izlucheniy na materialy. Moscow, 1960. Deystviye yadernykh izlucheniy na materialy (The effect of nuclear radiation on materials); doklady soveshchaniya. Moscow, Izd-vo AN SSSR, 1962, 219-234 TOPIC TAGS: copper, aluminum, magnesium, chromium steel, nickel containing steel, metal internal friction, metal fatigue, neutron irradiation, irradiation effect ABSTRACT: The internal friction (1/Q) and the normal elasticity modulus have been investigated in solution-heat-treated copper aluminum, and magnesium prior to and after irradiation at 80C with an integrated flux of 2.0^{1} x 10^{16} -5.0 x 10^{20} thermal n/cm2 (the number of fast neutrons with an energy of more than 1 Mev was 35%). 1/Q was measured at a stress of 2-20,000 g/mm². The plotted internal frictionstrain amplitude curves showed the existence of a critical strain (σ_{cr}) under which the 1/Q begins to be affected by the applied stress. The 1/Q and σ_{Cr} were found to be very sensitive to irradiation (see Fig. 1.). For example, the cr for irradiated copper increased 280 times and the minimum value of 1/Q decreased by two times compared with the initial value before irradiation. The changes in the value of 1/Q and **Card 1/3** 0102 0208

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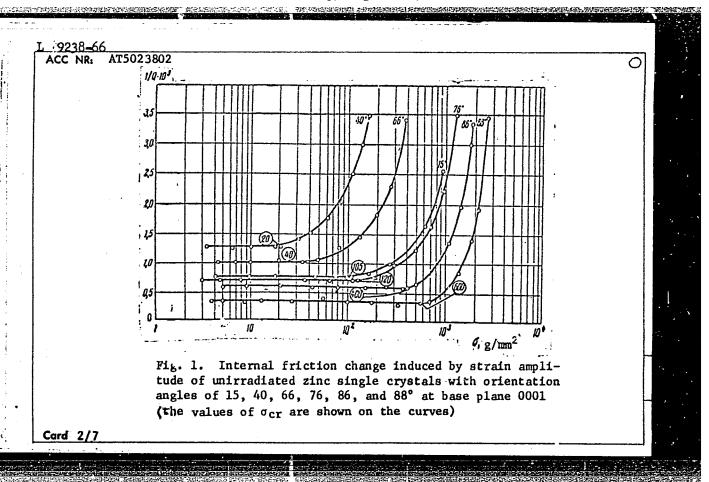
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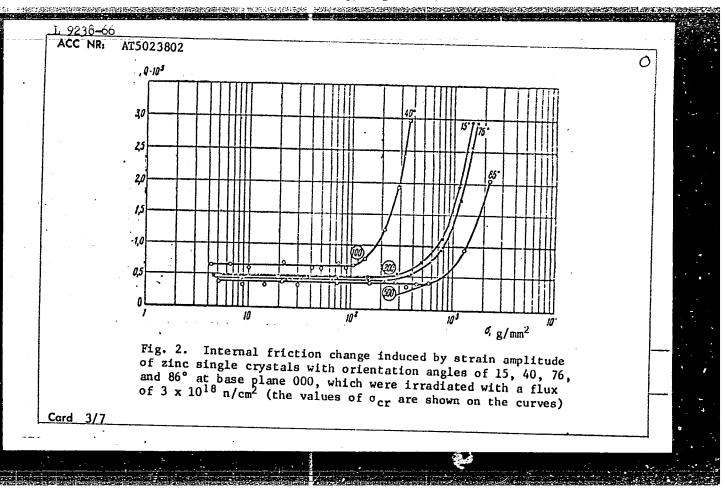
 $\sigma_{\rm cr}$ with irradiation doses equal to or less than $10^{17}~{\rm n/cm^2}$ are caused by the interaction of dislocations and point defects which resulted from elastic scattering of neutrons. In the case of plastic deformation of up to 27%, the point defects resulted from interaction between dislocations, and the increase in the value of 1/Q was considerably smaller. In distilled magnesium subjected to fatigue with a exclic stress of various amplitude before irradiation with an integrated flux of 1019 n/cm2 (thermal neutrons and about 10% fast neutrons with an energy above 1 Mev), the value of $\sigma_{\rm cr}$ was found to increase from the initial 5 g/mm² to 100 g/mm² after irradiation. In fatigue testing under a cyclic stress of 1600—4500 g/mm², distilled magnesium irradiated with an integrated flux of 1019 n/mp2 (thermal) had an endurance limit 10% higher than unirradiated magnesium. The effect of irradiation on the natural vibration frequency of specimens (the square of which determines the normal elasticity modulus) was investigated on irradiated copper and unirradiated 1Kh18N9T [AISI 321] stainless steel. The observed irradiationinduced behavior of the normal elasticity modulus can be explained by a manifestation of both the elastic and "nonelastic" properties of the metal, depending on the magnitude of the stress applied in dynamic measurement of the modulus. The "nonelastic" properties of the metal can be caused by migration of dislocations, while pure elastic properties manifest themselves only in the region of stresses $\sigma \leq \sigma_{cr}$.

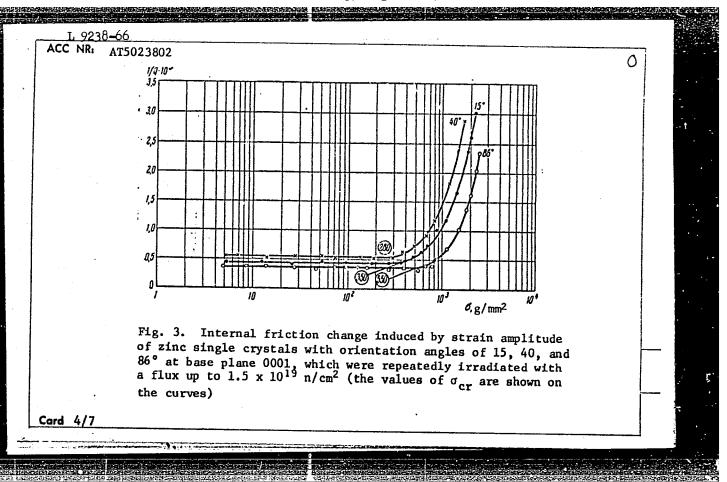
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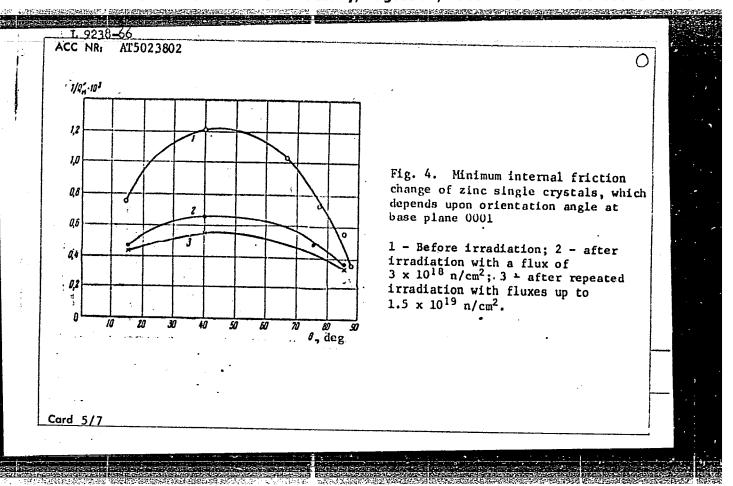


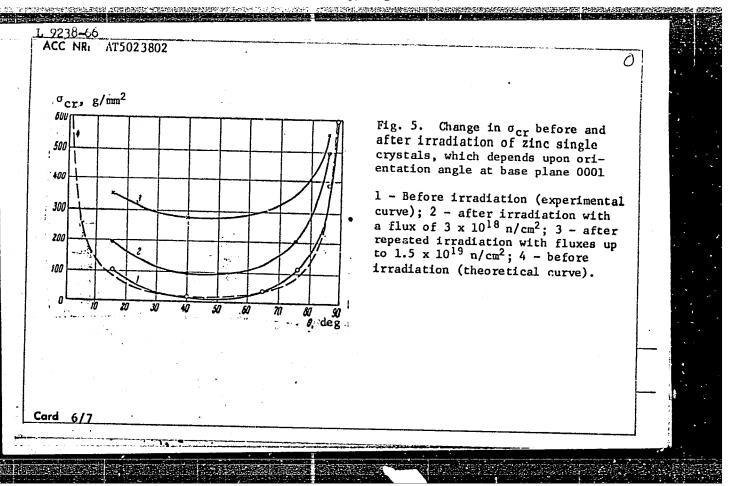
L 9238-66 EWT(d)/EWT(1)/EWT(m)/EWP(w)/EPF(n)-2/EWP(v)/T/EWP(t)/EWP(k)/EWP(b)/AT5023802 EWA(h)/EWA(c)/ETC(m) SOURCE CODE: UR/0000/62/000/000/0235/0241 JD/WW/EM/GG/GS Pravdyuk, N. F.; Pokrovskiy, Yu. I.; Vikhrov, V. I. AUTHOR: ORG: none B+1 TITLE: Effect of neutron irradiation on the internal friction of zinc monocrystals and polycrystals $\frac{1}{2}$ SOURCE: Soveshchaniye po probleme Deystviye yadernykh izlucheniy na materialy. Moscow, 1960. Deystviye yadernykh izlucheniy na materialy (The effect of nuclear radiation on materials); doklady soveshchaniya. Moscow, Izd-vo AN SSSR, 1962, 235-241 TOPIC TAGS: irradiation, neutron irradiation, zinc single crystal, zinc polycrystal, internal friction ABSTRACT: Zinc single crystals and polycrystals with various base plane angles and with orientation angles of 15, 46, 66, 76, 86, and 88° were irradiated with integrated fluxes of 3 x 10^{18} or 1.5 x 10^{19} n/cm², and the effect of irradiation on the internal friction was investigated. Results of investigations showing changes of internal friction, which are produced by the maximum strain amplitude (σ_{cr}) , at which the internal friction begins to depend upon it, in zinc single crystals and polycrystals with or without applying neutron irradiation are shown in Figs. 1-6. Card 1/7

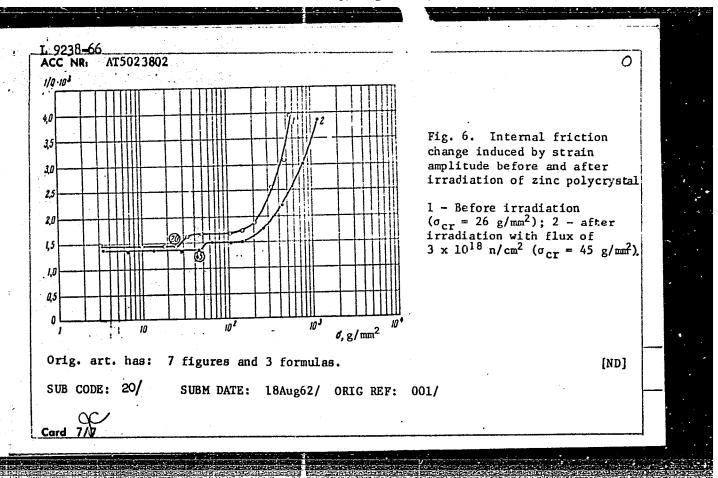












<u>L 53943-65</u> EWT(m)/EPF(c)/EPF(n)+2/EWG(m)/EPR/EWA(h) Pr-4, Ps-4/Pu-4 WW	TOTAL
ACCESSION NR: AT5013238	UR/3119/64/000/002/0051/0064 40	
AUTHOR: Pravdyuk, N. F.; Ivano Perevezentsev, V. N.	v, V. P.; Kuznetsov, V. N.; Vikhrov, V. I.;	
TITLE: Measurement of fast neuments on irradiation of materia	tron flows using threshold reactions during experi- Is in the RFT reactor	
SOURCE: AN LatSSR. Institut f Dozimetriya neytronov i gamma-l	iziki. Radiatsionnaya fizika, no. 2, 1964. uchey (Dosimetry of neutrons and gamma rays), 51-64	
TOPIC TAGS: fast neutron flux, flux measurement, reactor neutr	threshold reaction, neutron registration, neutron on flux, radiation dosimetry	
fluxes of fast neutrons using t	the problem of absolute measurements of integral he threshold reactions	
	[); $P^{31}(n, p) Si^{31}$; $S^{32}(n, p) P^{32}$; Co^{88} ; $Ci^{35}(n, \alpha) P^{32}$; $Ai^{27}(n, \alpha) Na^{54}$	
in the channels of the RFT reac	tor. The absolute isotope activity was measured by counter, a 4π slit scintillation counter, the β - γ -ters, and by calibrated proportional and crystal	
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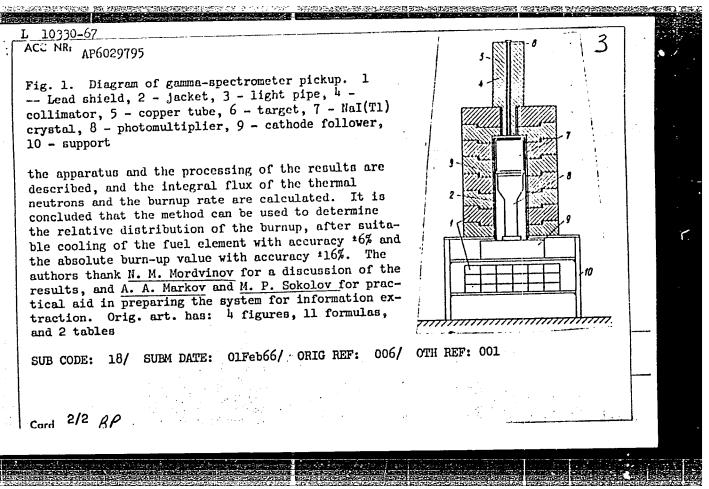
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counters with layers of fissionable material. 'he distribution of fast and thermal neutrons within the active zone was determined according to the relative method by means of fission counters and sulfur and copper activation. The article outlines the thoory of the experiments, describes the experimental method, and reports the results of the measurements. The methods used for the flux determin nation during the present investigation are quite complex and cumbersome and the success of any such measurement hinges to a great extent on the knowledge of the shape of the spectrum at various points within the reactor. Consequently, the authors suggest that the theoretically calculated spectra be used whenever possible. If the neutron spectrum for a point in the reactor is calculated reasonably accurately, a single threshold indicator is sufficient for its empirical normalization, If the accuracy of the calculation is in doubt, it can be checked and corrected by means of a series of threshold indicators (see, R. Nibson, Neutron dose monitoring for irradiation of materials in reactors. Contributions to the Harwell Symposium in December 1962, No SM 36/42; J. Moteff, Nucleonics, 20, 1962, 12, 56). "In conclusion, the authors thank Yu. G. Nikolayev and his co-workers for calculating the neutron spectra and for practical help in carrying out the experiments." Orig. art. has: 9 formulas, 5 figures, and 3 tables.

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L 10330-67 EWT(m)/EWP(t)/ETI 1JP(c) - 12-/68 CODE: UR/0089/66/021/002/0092/0096 ACC NRI AP6029795 (N) COUNCE CODE: UR/0089/66/021/002/0092/0096	
AUTHOR: Pravdyuk, N. F.; Vikhrov, V. I.; Pavlov, B. Yu.; Perevesontsov, V. N. 57	
ORG: none	•
TITLE: Determination of the burnup of the fuel element of the icebreaker "Lenin" from the Cs-137 activity without chemical separation	
SOUPCE: Atomnava energiya, v. 21, no. 2, 1966, 92-96	
TOPIC TAGS: reactor fuel element, cesium, uranium compound, enriched uranium, reactor neutron flux, gamma neutron reaction	
ABSTRACT: The authors determined the distribution of the burn-up along the length of ABSTRACT: The authors determined the distribution of the 0.66-Mev gamma lines of the Cs ¹³⁷ . the fuel element by measuring the intensity of the 0.66-Mev gamma lines of the Cs ¹³⁷ . in the reaction products with a scintillation, γ spectrometer with resolution 1012%. in the reaction products with a scintillation operated for The fuel element tested was made of uranium dioxide with 5.5% enrichment, operated for the fuel element tested was made of uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 11 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 12 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 12 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 12 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 12 places and two samples of the uranium dioxide were chosen cut in the hot chamber in 12 places and two samples of the uranium dioxide were chosen cut in the feet of the uranium dioxide were chosen cut in the feet of the uranium dioxid	
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PRAVDYUK, V.V.; kand. veyen: e-mershikh nauk; SCHCLOV, A.V.

Velocity lesses of ships under the effect of hydrometeorological conditions. Inform. abor. TSNIGHT no. 120. Sudevenh. i svinz' no. 27:42-47 '64. (HIRA 19:1)

Printing understand understive Fig.VDYUK, V.V., st. nauchn. setr.

[The Loran pulse-type radio navigation system] Impul's-nais radionavigusatonnania sistems "Loran." Nockva, Izd-vo "Transport," 1964. 131 p. (EHA 17:5)

1. TSentral'nyy nauchno-issledovatel'skiy institut morskogo flota (for Fravdyuk).

BIKOV, V.I., kand.tekhn.nauk; PRAVDYUK, V.V., kand.voyenno-morskikh nauk

Use of radio navigation systems (RNS) for navigation on approaches to harbors, on canals and in narrows. Trudy TSMIMF 3 no.47:
3-15 '63.

(MIRA 16:12)

DANILOV, L.; KAMENSKIY; SOKOLOV; PRAVDYUK, Ya.

Eliminating excessive load testing of bridge cranes Comments on an article by S.N.Ryzhov. Metallurg 10 no.4:31 Ap '65. (MERA 18:7)

1. Glavnyy mekhanik Cherepovetskogo metallurgicheskogo zavoda (for Danilov). 2. Ispolnyayushchiy obyazannosti glavnogo mekhanika Taganrogskogo metallurgicheskogo zavoda (for Kamenskiy). 3. Nachal'nik byuro tekhnicheskogo nadzoro Otdela glavnogo mekhanika Taganrogskogo metallurgicheskogo zavoda (for Sokolov). 4. Glavnyy mekhanik Krivorozhskogo metallurgicheskogo zavoda (for Pravdyuk).

"APPROVED FOR RELEASE: Tuesday, August 01, 2000

CIA-RDP86-00513R001342

SOURCE CODE: UR/0185/66/011/005/0563/0565 EWT (m) L 08936-67 ACC NR: AP6016050 AUTHOR: Korzh, I. O.; Mishchenko, V. O.; Pravdyvyy, M. M.; Prykhod'ko, V. P.; 50 Sklyar, M. T.; Totakyy, I. A. ORG: Institute of Physics, AN UkrSSR, Kiov (Instytut fizyky AN UkrSSR) TITLE: Measurement of angular distribution of neutrons with energies of 0.3, 0.5, and C.8 Mev in elastic scattering on titanium and cobalt nuclei SOURCE: Ukrayins'kyy fizychnyy zhurnal, v. 11, no. 5, 1966, 563-565 TOPIC TAGS: angular distribution, elastic scattering, neutron scattering, scattering cross section, titanium, cobalt, nuclear energy level, nucleus ABSTRACT: These measurements were carried out because the available data on angular distribution in elastic scattering of neutrons with energies of less than 1 Mev are inadequate for calculating the mean nuclear physical constants with sufficient accuracy. The measurement results are given in the accompanying table from which it is seen that the data on the total cross section obtained by calculation are in good agreement with the experimental data (column 3) obtained by D. Hughes and J. Harvey (Neutron Gross Section, Second Edition, ENL-325, 1958). Orig. art. has: 2 formulas, 2 figures, and 1 table. Card 1/2

	•.						
Element	Energy of neutron from photoneutron source E _n ,	Total cross section ot, barn		Elastic scattering cross section o e, barn	Сов 🖯	Transport cross section at elastic scattering of tre, barn	
Ti	0.3 0.5	2.79 2.42	2.85 2.72	2.69±0.19 2.57±0.10	0.14 <u>†</u> 0.02 0.17 <u>†</u> 0.01	2.30±0.22 2.12±0.12	
Co	0.5 0.8	4.48 3.42	3.54	4.77±0.24 3.73±0.26	0.13 <u>+</u> 0.01 0.21 <u>+</u> 0.03	4.14±0.27 2.94±0.32	
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	SUB CODE:	20/ SUBM	DAME: 307	an66/ ORIG R	EF: 003/	OTH REF: 004	

KREN, Emil; PRAVECZKI, Endre

Classifying magnetic structures and the method for determining the neutron diffraction. Magy fix folyoir 12 no.4:387-414 '64.

1. Solid Physics Laboratory, Central Research Institute of Physics, Hungarian Academy of Sciences, Budapest.

PRIVECZKI, Endre

Electric resistance of ferromagnetic substances. Koz jiz kozl

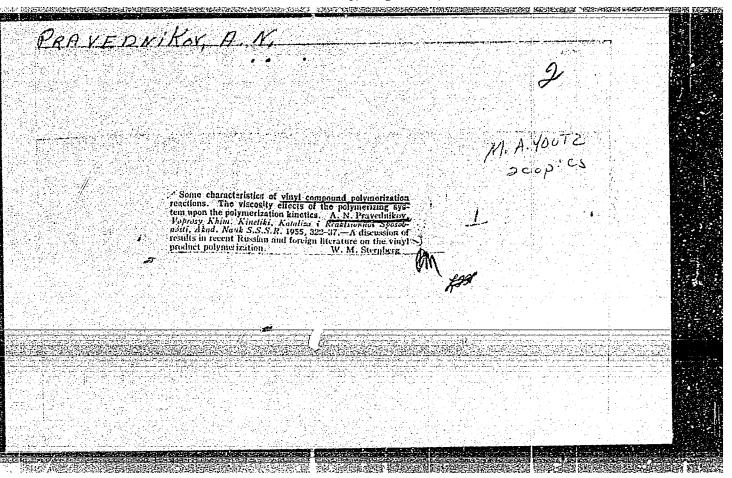
MTA 13 no.1:27-31 '65.

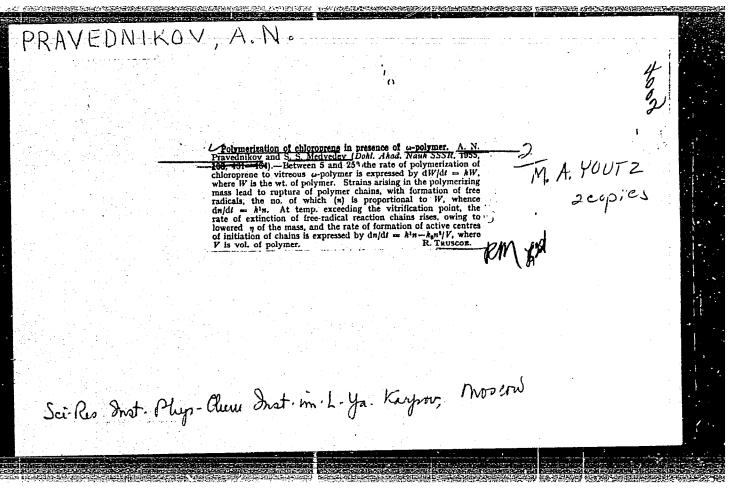
1. Submitted November 3, 1964.

PRAVEDNIKOV, A. N.

Diesertation: "Some Characteristics of the Kinetics of Extensive Polymerization of Vinyl Compounds." Cand Chem Sci, Sci Res Physicochenical Inst inent L. Ya. Karpov, Moscow, 1953. (Referatively Zhurnal--Kalmiye, Moscow, No 5, Mar 54)

S0: Sum 243, 19 Oct 54





FRAVEDNIKOV, A.M.; MEDVEDEV, S.S.

Study of the chloroprene \(\text{-polymerization mechanism by means of labelled atoms. Dokl. AN SSSR 109 no.3:579-581 J1 '56.

(KIRA 9:10)

1. Chlen-korrespondent Akademii nauk SSSR (*- Modwedev)

(Chloroprene) (Carbon--Isotopes)

"APPROVED FOR RELEASE: Tuesday, August 01, 2000

CIA-RDP86-00513R001342

F.

PRAVEDNIKOV, A.N.

USSR/ Chemistry of High-Molecular Substances

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11943

Author : Pravednikov A.N.

Inst : Academy of Sciences USSR

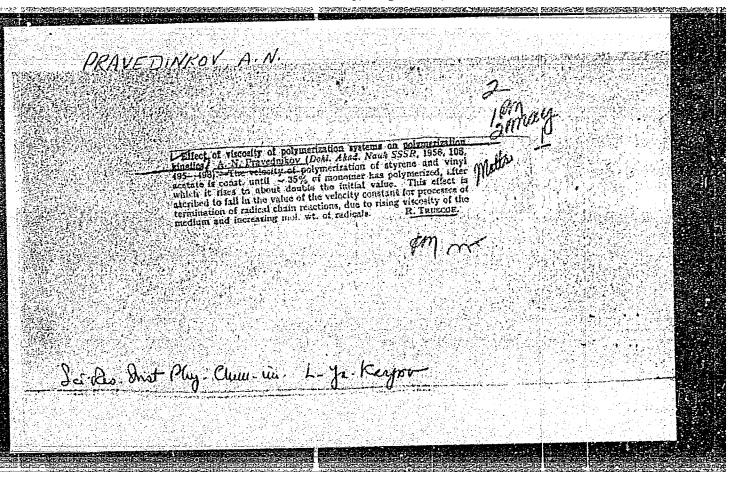
Title : Effect of Viscosity of the Polymerization System on

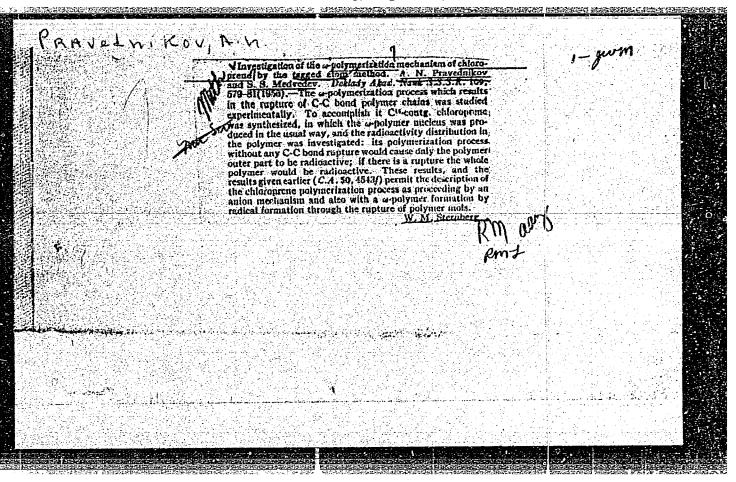
Polymerization Kinetics

Orig Pub : Dokl. AN SSSR, 1956, 108, No 2, 495-498

Abstract : See also RZhKhim, 1956, 22623

Card 1/1





PRAVEDUINOV, A. H., and WEDVEDLY, S. S.

"Kinetics and medianism of chloroprene polymerization," a paper presented at the 9th Congress on the Chemistry and Physics of High Polymers, 28 Jan-2 Feb 57, MOscow, KarpovInst.

B-3,084,395

5(4) AUTHORS: SOV/20-122-2-24/42

Ying Shen-K'ang, Pravednikov, A. N., Medvedev, S. S., Member,

Academy of Sciences, USSR

On the Mechanism of the Cross Linkage of Polymer Chains Under TITLE:

Gamma Radiation (O mekhanizme sshivaniya the Action of

polimernykh tsepey pod deystviyem gamma-izlucheniya)

Doklady Akademii nauk SSSR, 1958, Vol 122, Nr 2, pp 254-257 PERIODICAL:

(USSR)

The formation of transverse bonds must be connected with ABSTRACT:

secondary processes in which radicals take part. The velocity of the cross linkage of the chains is constant with respect to time and proportional to the intensity of the radiation. The most simple assumption concerning the mechanism of the cross linkage is the following one: During radiolysis, the transverse bonds are formed by recombination of the polymer radicals to double bonds of the polymer molecules. The steady state with respect to the concentrations of the double bonds and free radicals will be not reached. The recombination of

the radicals and the joining-together of the radicals to

double bonds play only an unimportant rôle. In order to under-Card 1/4

501/20-122-2-24/42

On the Mechanism of the Cross Linkage of Polymer Chains Under the Action of Gamma Radiation

stand the mechanism of the processes which lead to the formation of transverse bonds, the radical reactions in the irradiated polymer (especially the reactions in which atomic hydrogen takes part) have to be investigated. This hydrogen atom may either loose its excess energy or react according to one of the following reactions: 1) interaction with an other hydrogen atom: $H + H \longrightarrow H_2$, 2) interaction with free radicals produced during the irradiation

$$\sim \text{CH}_2 - \dot{\text{CH}} - \text{CH}_2 \sim + \text{H} \sim \text{CH}_2 - \text{CH}_2 - \text{CH}_2 \sim + \text{H}_2$$

3) interaction with the double bonds: $\sim \text{CH}_2\text{-CH=CH} \sim + \text{H} \longrightarrow \text{CH}_2\text{-CH-CH}_2 \sim \cdot$ 4) Detachment of the hydrogen atom from the polymer molecule: $\sim \text{CH}_2\text{-CH}_2\text{-CH}_2 \sim + \text{H} \longrightarrow \sim \text{CH}_2\text{-CH-CH}_2 \sim + \text{H}_2$. The velocities of these reactions mainly depend on the probability of the collision of the hydrogen atom with the various

Card 2/4

SOV/20-122-2-24/42

On the Mechanism of the Cross Linkage of Polymer Chains Under the Action of Gamma Radiation

groupings. Below the vitrification temperature. T, the velocities of the cross linkage of polyethylene and polyvinylchloride do not depend on the temperature. In polymers which contain a sufficiently high number of lateral groupings (bokovaya grupirovka), the "cold" hydrogen atoms will take part in the reactions of cross linkage also below the vitrification temperature. In order to prove this assumption, the authors investigated the influence of the γ-radiation on polymers of

the structure $\begin{pmatrix} -C_3^H 6^{-CH}_{1}^{-CH} 2^{-} \end{pmatrix}$

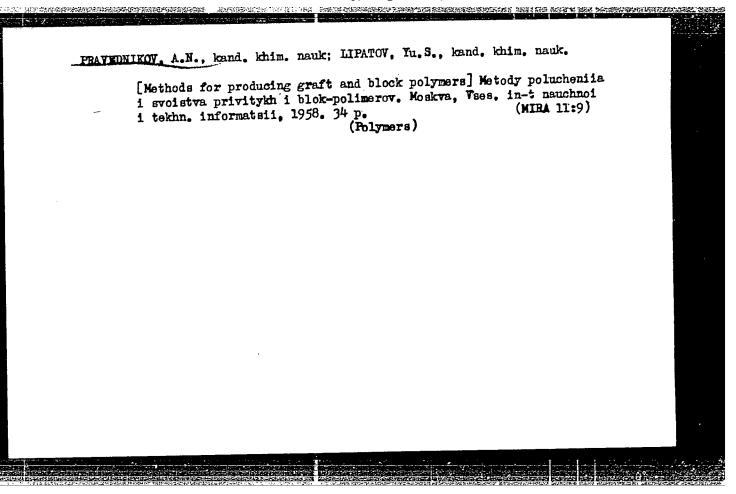
There are 2 figures, 3 tables, and 2 references, 0 of which is Soviet.

ASSOCIATION: Nauchno-issledovatel'skiy fiziko-khimicheskiy institut im.

L. Ya. Karpova

(Physical-Chemical Scientific Research Institute imeni L. Ya.

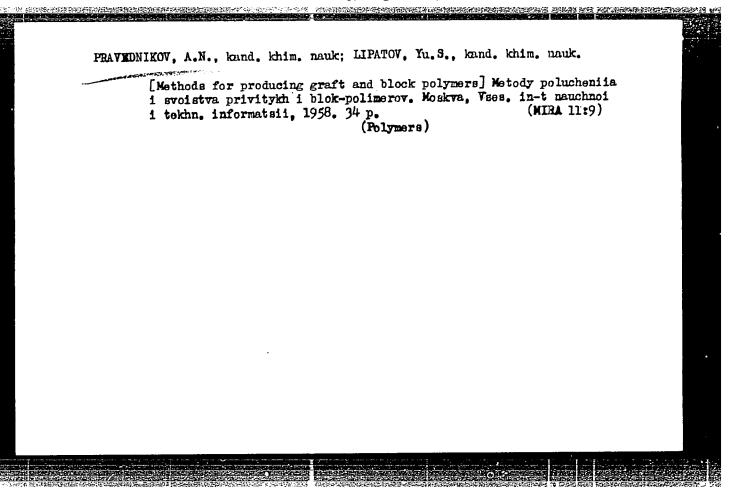
Card 3/4 Karpov)



PRAVEDNIKOV, A. N.. YING SHENG KANG, and MEDVEDEV, S. S.

"On the Mechanism of Cross-Linking of Polymer Chains Under Gamma-Irradiation."

paper to be presented at 2nd UN Intl. Conf. on the peaceful uses of Atomic Energy, Geneva, 1 - 13 Sept 58.



BUGAYENKO, L.T.; NIKITINA, T.S.; PRAVEDNIKOV, A.N.; MALINSKIY, Yu.M.

[Chemical action of ionizing radiation] Knimicheskoe deistvie ionizirulushchikh izluchenii. Moskva, 1958. 84 p. (MIRA 12:4)

(Radiochemistry)

5(3), 5(4)

AUTHORS:

Aleksandrova, Yu. A., Huang Yu-li , SOV/20-123-6-20/50

Pravednikov, A. N., Medvedev, S. S. Academician

TITLE:

Reactions of Oxygen-Containing Radicals of the RO'Type

(Reaktsii kislorodsoderzhashchikh radikalov tipa RC)

PERIODICAL:

Doklady Akademii nauk SSSR, 1958, Vol 123, Nr 6,

pp 1029 - 1032 (USSR)

ABSTRACT:

The reactions under review were carried out mainly on model systems in which the RO-radicals were formed from the decomposition of dialkyl peroxides. The authors found that at the decomposition of ditertiary-butyl-peroxide, dissolved in hydrocarbons, in addition to methane at 195° the resulting amount of acetone is about 12 times that of tertiary butyl alcohol. This is indicative of a higher activation energy than had been found by J. H. T. Brook (Bruk) (Ref 2). Proceeding from scheme (I), (II), (III), nearly all tertiary butoxy radicals are likely decompose under cleavage of the C-C bond as can be assumed from the results obtained. This is, however, in contradiction to the data published on the "thermal-oxidative" destruction of the carbon chain polymers (Ref 4). It

Card 1/3

21(8) PHASE I BOOK EXPLOITATION

sov/2326

Bugayenko, L. T., T.S. Nikitina, A. N. Pravednikov, and Yu M. Malinskiy

Khimicheskoye deystviye ioniziruyushchikh izlucheniy (Chemical Action of Ionizing Radiation) Moscow, 1958. 84 p. (Series: Khimicheskaya promyshlennost') Errata slip inserted. 1,500 copies printed.

Sponsoring Agencies: USSR. Gosudarstvennyy nauchno-tekhnicheskiy komitet, and Akademiya nauk SSSR. Vsesoyuznyy institut nauchnoy i tekhnicheskoy informatsii. No contributors mentioned.

FURPOSE: The book is intended for chemists and chemical engineers.

COVERAGE: The book discusses the effect of ionizing radiation on various chemical processes. The effect of radiation on inorganic and organic compounds, on polymerization in the liquid, gaseous and solid phases, and on the properties of polymers is adequately covered. No personlities are mentioned. There are 495 references: 67 Soviet, 343 English, 16 German, 66 French, and 3 Italian.

Card 1/3

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lectrochemical properties of irradiated solutions	19	
to a selection and doctmet we		
ADS OI LEGISTION STOT GOSTING OLD	20	
ffect of radiation on solids	22	
adiation reactions in the gaseous phase	23	
	24	
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inturated hydrocarbons		·
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0/2		
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Chemical Action of Ionizing (Cont.)	SOV/2326	
Oxygen compounds	34	
Halogen compounds	34 36 38 46	
Aromatic compounds	38	
Bibliography	46	
Radiation by Polymerization	51	
Polymerization in the liquid phase	52	
Polymerization in the gaseous phase	52 57 58 60	
Polymerization in the solid phase	50	
Bibliography	00	
Effect of Ionizing Radiation on Polymers	62	
Adhesion of polymers by radiation	65	
Degradation of polymers by radiation	73	
Effect of ionizing radiation on solutions of polymers	79	
Bibliography	82	
Conclusion	86	
AVAILABLE: Library of Congress		
	TM/fel	
Card 3/3	9 - 9 - 59	

Problems in Physical Chemistry (Cont.) SOV/4386	
Rreger, A. Kh., M. A. Dembrovskiy, L. A. Dmitriyev, L. L. Sunitsa and Yu. S. Ryabukhin. Study of the Field of Forces of Dosages From a Cylindrical Irradiator With Co ⁶⁰ as a Powerful Source of Y Radiation	132
Potapov, V. K., B. G. Vasil'yev and N. N. Tunitskiy. Study of the Ionization and Dissociation of n-Octane and n-Nonane Molecules by the Method of Bombardment With "Quasi-Monokinetic" Electrons	146
Baberkin, A. S. Radiation-Chemical Effects in Solid Inorganic Salts	163
Maksimov, M. P., A. V. Zimin, and R. V. Dzhagatspanyan. Radiation-Chemical Chlorination of Benzene	169
Froskurnin, M. A., Ye. V. Barelko, and L. I. Kartasheva. Course of the Process of Benzene Oxidation in an Aqueous Solution Under the Action of Radiation	177
Card 4/5	

	्रेक्टा दे <mark>र</mark> िकेश स्थापनी देश सम्बद्ध है है।	
Problems in Physical Chemistry (Cont.) SOV/4386		
Kucera, J (Czechoslovakia), Ye. V. Barelko, L. I. Kartasheva, P. N. Komarov, and M. A. Proskurnin. Decomposition Products of Phenol Formed During the Radiolysis of Benzene in an Aqueous Solution	183	
Sharpatyy, V. A., and G. A. Gol'der. The Problem of the Phase Composition of the System H ₂ O-NaNO ₃ -NaOH at Low Temperatures	189	
Orekhov, V. D., and A. A. Zansokhova. Sensitization of the Radiolytic Oxidation of Leucoform Dyes	194	
AVAILABLE: Library of Congress		
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BAGDASAR'YAN, Khristofor Stepanovich. Prinimal uchastiye: PRAVEDNIKOV, A.N.. MEDVEDEV, S.S., otv.red., akademik; BANKVITSER, A.L., red.izd-va; KUZ'NIN, I.F., tekhn.red.

[Theory of polymerization by radicals] Teoriia radikal noi polimerizatsii. Moskva, Izd-vo Akad.nauk SSSR, 1959. 297 p.
(MIRA 12:7)
(Radicals)

YUR'YEV, V.M.; PRAVEDNIKOV, A.N.; MEDVEDEV, S.S., akademik

Effect of side chains on the rate of oxidation of carbon chain polymers. Dokl.AN SSSR 124 no.2:335-337 Ja '59.

(MIRA 12:1)

1. Nauchno-issledovatel'skiy fiziko-khimicheskiy institut imeni L.Ya. Karpova.

(Oxidation) (Polymers)

Reactions of Oxygen-Containing Radicals of the RO Type SOV/20-123-6-20/50

can be concluded from the results that acetone here is not only formed as a consequence of the reaction :

 $(CH_3)_3CO$. $\xrightarrow{k_1} (CH_3)_2C \longrightarrow O+.CH_3$ (I), but also in consequence of some other reaction the velocity of which considerably depends on temperature. Such a reaction can be that of the RO-radicals with one another. At low temperatures the concentration of the RO. radicals is low and the reaction proceeds slowly (Ref 2). In order to prove the acceleration of this reaction at increasing temperature or at a considerable increase in concentration of the peroxide, the authors have investigated the decomposition of the di-tertiary-butyl-peroxide in an isopropyl-benzene solution at 120 - 1500 and in the concentration range from 4 up to 16 percentage by weight. Figure 1 shows that the ratio of the concentrations of acetone (a) and tertiary butyl alcohol (b) a/b increases with an increasing concentration of the peroxide. Therefore the reaction order of the formation of these compounds with respect to the peroxide concents tion is not equal to 1. According to various computations the authors conclude that the acetone

Card 2/3

Reactions of Oxygen-Containing Radicals of the RC Type SOV/20-123-6-20/50

formation under the above conditions at temperatures of about 200° is largely related with the bimolecular reaction:

(CH₃)₃CO·+(CH₃)₃CO· (CH₃)₂C = O+CH₃-O-C(CH₃)₃ (IV) and not with the monomolecular decomposition of the RO· radicals. In the case of high-polymers the reaction (IV) must lead to a rapid variation of the distribution regarding the molecular weights. This occurs indeed in the radical stages of the polyethylene oxidation. This variation is accompanied by the occurrence of ether bridges between the macromolecules. There are 4 figures, 1 table, and 5 references, 3 of which are Soviet.

ASSOCIATION:

Nauchno-issledovatel'skiy fiziko-khimicheskiy institut im. L. Ya. Karpova (Scientific Physical-Chemical Research Institute

imeni L. Ya. Karpov)

SUBMITTED:

September 29, 1958

Card 3/3

5(3)

AUTHORS: Yur'yev, V. M., Pravednikov, A. N.,

SOV/20-124-2-26/71

DESCRIPTION OF THE PROPERTY OF

Medvedev, S. S., Academician

TITLE:

Influence of Side Chains on the Rate of Oxidation of Carbon Chain Polymers (Vliyaniye bokovykh otvetvleniy na skorost' okisleniya karbotsepnykh polimerov)

PERIODICAL:

Doklady Akademii nauk SSSR, 1959, Vol 124, Nr 2, pp 335-337 (USSR)

ABSTRACT:

The principal reactions in the oxidation of hydrocarbons are the following:

 $R \cdot + 0_2 \longrightarrow ROO \cdot (1); ROO \cdot + RH \longrightarrow ROOH + R \cdot (2);$

ROOH \longrightarrow RO· + OH (3); RO· + RH \longrightarrow ROH + R· ,

•OH + RH \longrightarrow H₂O + R• (4) . The rates of all these elementary

reactions determine the rate of oxidation. As is known the rate is considerably decreased on the transition from low molecular weight to high molecular weight compounds of analogous structure (Refs 1, 2). This might be explained as follows: The removal of one hydrogen atom from the hydrocarbon atom is accompanied by a

Card 1/3

THE PROPERTY OF THE PROPERTY O

Influence of Side Chains on the Rate of Oxidation of Carbon Chain Polymers

SOV/20 -124-2-26/71

transition of the corresponding link of the molecule from a tetrahedral to a plane configuration. In polymers, links of the polymer chain are displaced. This is bound to increase the activation energy and thus to retard the reaction (as compared with the analogous reactions of low molecular weight compounds). The separation of one hydrogen atom from a side group (methyl-, propyl- and others) is not accompanied by a displacement of the links of the polymer chains and must possess the same activation energy as the corresponding reactions of the low molecular weight compounds. It can therefore be expected that the oxidation of the polymers with comparatively short side chains will take place mainly on the side chains. To control this assumption the authors synthesized polymethylene as well as polymers which contained the methyl and propyl side groups (Ref 4). The experiments concerning the oxidation of these polymers have shown that the introduction of side groups rapidly increases the absorption rate of oxygen (Fig 1,a); at the same time the number of oxygen molecules which are used for the cleavage of the principal chain (Figs 3, 4) increases, i.e. the oxidation really proceeds in the side chains prevalently. At a high oxidation intensity of the polymers which were produced by decomposition

Card 2/3

Influence of Side Chains on the Rate of Oxidation of Carbon Chain Polymers

SOV/20-124-2-26/71

of the diazo compounds, a "sewing up" (zashivaniye) of the polymer results as a consequence of ether bridges between the macro-molecules. A very low molecular fraction appears within the system as well. Possibly, these variations are due to the proceeding of a bimolecular reaction under participation of 2 oxygen containing radicals (Ref 6). Polystyrene is not "sewed up" at an oxidation intensity of up to about 20 ml 02 per 1 g polymer, since the concentration of the radicals and the oxidation rates, respectively, seem to be too low. There are 4 figures and 6 references, 3 of which are Soviet.

ASSOCIATION:

Nauchno-issledovatel'skiy fiziko-khimicheskiy institut im. L. Ya. Karpova (Scientific Physical and Chemical Research

Institute imeni L. Ya. Karpov)

SUBMITTED:

September 29, 1958

Card 3/3

PRAVEDNIKOV, A.N.

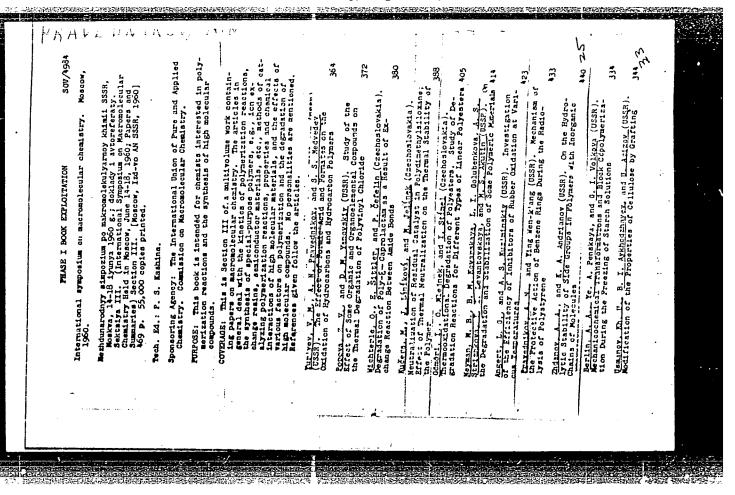
Effect of polar and steric factors on the reactions of chain transfer and chain growth in polymerization. Probl. fiz.khim. no.2:5-13 59. (MIRA 13:7)

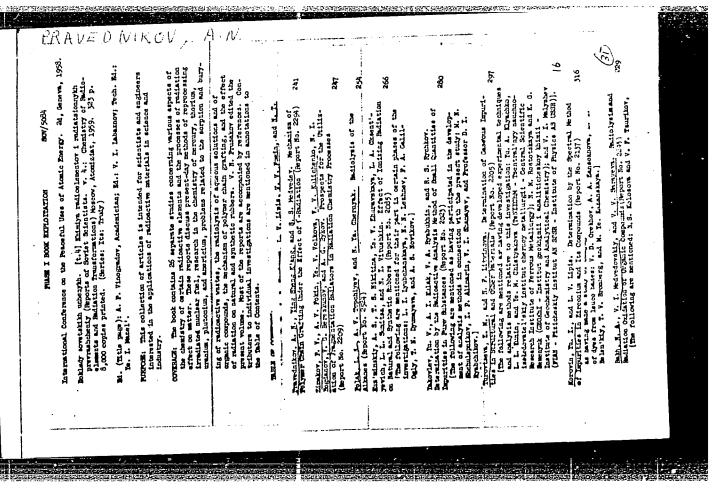
1. Laboratoriya polimerizatsionnykh protsessov Mauchno-issledovatel'skogo fiziko-khimicheskogo instituta imeni L.Ya. Karpova.

A CONTROL OF THE STREET OF THE

(Polymerization) (Vinyl compounds)

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REMEMBER, A., ROMNITCHIN, N. FRANCESCON, M. A., and TRUCKY, A. MAILTHAIN SECRETIFIE Begarch Institute of Synthetic Rubber Local S. V. Lebedar, Lethington - "Sature of solrendar-wight distribution and properties of styrene-butchies mibers depending on polymerization conditions" (Group 3-A) RAVEHINDY A. N. TELEBRY, V. H., T. L., Reprise-General Institute 1984 i. N. Kaper, Assoc - Threstitations containing quaternary atoms of carbon" (Group below Market and Carbons of the Acadery of Sciences USSA, Laningrad - "Stereo-regularity and optical anisotropy of accroniscules" (Group or specified) THEORY, Enabledy, and ENIESZYCH, T. L., Candery of Sciences USSA, Tablent, Ubbedistan - "The trestitation of the cotton callinose polydispersity scording to the solecular weight" (Group out specified) THEORY A. S., Institute of thesical Revises of the Acadery of Sciences USSA, Moscov - "On the Minetics of forerlately polyseriarion and polyroralization of forerlately of J-2)	To be submitted for the Exercational Symposium on Macromolecular Chemistry, Mactivel, Caracle, 77 Jul -1 Lag 1961. EXECUTION IN. Institute of High Molecular Chemistry, Audiency of Sciences USSR, Marzon, N., Dark threenity, Durhan, H.C "Enacticity of the Intitude, H.C "Enacticity of the Intitude of Fine Control (Group 2) 2004/Hill. 2014 A. and EXECUTION. Notice of the Control (Group 2) 2004/Hill. 2014 A. and Exercises of Group 2) 2004/Hill. 2014 A. and Exercises of Fine Control (Group 2) 2 2004/Hill. 2014 A. and Exercise of Fine Control (Group 3) 3 2004/Hill. 2014 A. Exercise Lean I. The Exercise Department of Mg crystal Structures in Polymeria and User properties (Group 3) 2 1 Institute of Exercises (Group 3) 2 1 Institute of Exercises (Group 3) 3 2 1 Institute of Exercises (Group 3) 3 2 1 Institute of Exercise (Group 3) 3 2 1 Institute of Exercise Department of McCaracle, A. A. Exercise Department of McCaracle, A. Exercise Department of McCaracle, A. Exercise Department of McCaracle,	





APPROVED FOR RELEASE: Tuesday, August 01, 2000

CIA-RDP86-00513R0013429

5(3) \$00\/20-127-3-33/71

AUTHORS: In Shen-kan, Pravednikov, A. N., Medvedev, S. S., Academician

COLUMN TERMINATURE SERVICE DE LA SESSE DESERVICIONES DE LA SESSE D

TITLE: The Mechanism of the Screening Effect of Benzene Rings in

the Hydrolysis of Polystyrene

PERIODICAL: Doklady Akademii nauk SSSR, 1959, Vol 127, Nr 3, pp 595-598

(USSR)

ABSTRACT: Polystyrene has a considerable radiation stability compared

to polyethylene and compounds containing no phenyl group. For cross-linking an exceedingly high energy is necessary. In this connection, the reaction is investigated which takes place during irradiation on polystyrene. The reaction mechan-

ism may be represented in the following manner:

Card 1/4

SOV/20-127-3-33/71
The Mechanism of the Screening Effect of Benzene Rings in the Eydrolysis of Polystyrene

Card 2/4

SOV/20-127-3-33/71

The Mechanism of the Screening Effect of Benzene Rings in the Hydrolysis of Polystyrene

A hydrogen atom breaks loose from the aliphatic chain; it may react with the styrene by a further separation of a hydrogen atom (2), or it may link up to the benzene ring by forming a free cyclohexadienyl radical (3). The latter reaction develops very rapidly. If, further, (II) reacts with (II), a"transversal compound" may be formed (4), or (II) reacts with (III), in which case this reaction may lead to the "transversal compound" (5), and with further disproportionation (6) to the re-formation of the benzene ring. Experiments

Card 3/4

SOV/20-127-3-33/71

The Mechanism of the Screening Effect of Benzene Rings in the Hydrolysis of Polystyrene

carried out with deuterium-substituted toluene proved the mechanism mentioned in (6). Data hereon (deuteron content in the compounds obtained and energy used for transition of a D-atom into the benzene ring) are given in table 1. The high stability of polystyrene to cross-linking may be explained by the disproportionation of the primary radicals with the cyclohexadiene radical. There is 1 table and 7 English references.

ASSOCIATION: Nauchno-issledovatel'skiy fiziko-khimicheskiy institut im.

L. Ya. Karpova

(Scientific Research Institute for Physical Chemistry imeni

L. Ya. Karpov)

SUBMITTED: April 17, 1959

Card 4/4

Prince Enchary A.A.

PHASE I BOOK EXPLOITATION SOV/4386

Moscow. Fiziko-khimicheskiy institut

Problemy fizicheskoy khimii; trudy, vyp. 2 (Problems in Physical Chemistry; Transactions of the Institute, no. 2). Moscow, Goskhimizdat, 1959. 202 p. 1,000 copies printed.

Editorial Board: Ya. M. Varshavskiy, Doctor of Chemical Sciences;
G. S. Zhdanov, Doctor of Chemical Sciences; V. A. Kargin,
Academician; Ya. M. Kolotyrkin, Doctor of Chemical Sciences
(Resp. Ed.); S. S. Medvedev, Academician; S. Ya. Pshenzhetskiy,
Doctor of Chemical Sciences; V. M. Cherednichenko, Candidate
of Chemical Sciences; V. S. Chesalova (Editorial Secretary),
Candidate of Chemical Sciences; Ed.: I. A. Myasnikov; Tech.
Ed.: Ye. G. Shpak.

PURPOSE: This collection of articles is intended for physical chemists.

COVERAGE: The collection is the second issue of the Transactions of the Scientific Research Institute of Physical Chemistry imeni L. Ya. Karpov. It contains 17 articles which review Card 1/5

24667

3/081/61/000/009/014/015 B101/B205

Ying Sheng-k'ang, Pravednikov, A. N., Kolmanson, A. E.

TITLE:

Investigation of $\gamma\text{-}\textsc{irradiated}$ vinyl polymers by means of

electron paramagnetic resonance

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 9, 1961, 645, abstract

9P85 (9R85) ("Huahsüeh Hsüehpao, Acta chim. sinica", 1960,

26, no.3, 164-168)

TEXT: On radiolysis of polyvinyl chloride (I) and polynitrile acrylic acid, the concentration of free radicals determined by means of epr decreases with rising temperature. (Irradiation was performed below vitrification temperature). The yield of free radicals obtained by radiolysis of (I) is higher than that obtained by radiolysis of polyethylene, whereas the yield of cross links is lower. The authors believe that electrons produced by radiolysis of (I) can be captured by polymer molecules. [Abstracter's note: Complete translation.]

Card 1/1

21,668

s/081/61/000/009/015/015 B101/B205

5.4600

Ying Sheng-k'ang, Teleshov, E. N., Pravednikov, A. N., AUTHORS:

Medvedev, S. S.

Mechanism of the radiolysis of polyisobutylene TITLE:

Referativnyy zhurnal. Khimiya, no. 9, 1961, 645, abstract PERIODICAL:

9P86 (9R86) ("Hua-hsüeh Hsüeh-pao, Acta chim. sinica", 1960,

26, no.3, 157-163)

TEXT: The temperature dependence of the rate of destruction and cross linking in the radiolysis of polyisobutylene shows the same character. This is ascribed to the fact that secondary radical reactions play a significant role in the destruction of the polymer. The rate of destruction is slowed down when the polymer is in a swollen state. If swelling takes place in toluene tagged with C^{14} , activity will appear in the polymer after irradiation, while the intrinsic viscosity remains unchanged. The authors believe that this phenomenon is related with the retardation of the reaction of bimolecular disproportionation. Radiolysis of a mixture of low-molecular and high-molecular fractions of

Card 1/2

21,668

S/081/61/000/009/015/015 B101/B205

Mechanism of the radiolysis of ...

polyisobutylene (low-molecular fraction tagged with C¹⁴) causes activity in the high-molecular fraction, too. From this, it is concluded that a bimolecular mechanism might underlie the destruction of the polymer. [Abstracter's note: Complete translation.]

Card 2/2

BEMFORD, K.[Bamford, C.H.]; BARB, U.[Barb, W.G.]; DZHENKINS, A.

[Jenkins, A.D.]; ON'ON, P.[Onyon, F.F.]; GRITSENKO, T.M.,

kand.khim. nank, [translator]; MILYUTINSKAYA, R.I., kand.

khim. nauk, [translator]; PRAVEDNIKOV, A.N., kand. khim.

nauk [translator]; MALINSKIY, Yu.M., kand. khim. nauk, red.;

KHODETSKAYA, Z.F., red.; PRIDANTSEVA, S.V., tekhn. red.

[Kinetics of vinyl polymerization by radical mechanisms] Kinetika radikal'noi polimerizatsii vinilovykh soedinenii. [By] C.H. Bamford i dr. Moskva, Izd-vo inostr. lit-ry, 1961. 345 p. Translated from the English. (MIRA 15:3)

(Vinyl compound polymers) (Radicals (Chemistry))

PRAVEDNIKOV, A.N.; KARDASH, I.Ye.; BAZOV, V.P.; YELISEYEVA, N.V.; SHARPATYY, V.A.; MEDVEDEV, S.S., akademik

Free-radical polymerization of triazine cycles. Dokl. AN SSSR 151 no.6:1347-1349 Ag '63. (MIRA 16:10)

I. 18897-63 EPR/EPF(c)/EWP(j)/EWT(m)/BDS ASD Ps-4/Pr-4/Pc-4 RM/WW/ACCESSION NR: AP3006596 S/0020/63/151/006/1347/1349 MAY/JFW

AUTHORS: Prayednikov, A. N.; Kardash, I. Ye.; Bazov, V. P.; Yeliseyeva, N. V.; Sharpaty*y, V. A.; Medvedev, S. S. (Academician)

TITLE: Free-radical polymerization of triazine bycles

SOURCE: AN SSSR. Doklady*, v. 151, no. 6, 1963, 1347-1349

TOPIC TAGS: free radical, polymerization, triazine, triazine cycle, free-radical polymerization

ABSTRACT: The present article reports the results of spectroscopic and electron paramagnetic resonance analysis of the polymers obtained by heating triazines with perfluoracetone as a source of CF₃ radicals/at 520C. The free-radical polymerization of triazine cycles, evidently representing addition of the free radical to the cycle on the double bond with subsequent opening of the cycle, must be accompanied at high temperatures by depolymerization, by a splitting of the monomeric by a unit from the polymeric radical. Orig. art. has: 1 formula 2 figures.

ASSOCIATION: none SUBMITTED: 28May63 SUB CODE: CH

Card 1/1

DATE ACQ: 27Sep63 NO REF SOV: 000 ENCL: 00 OTHER: 000

APPROVED FOR RELEASE: Tuesday, August 01, 2000

CIA-RDP86-00513R0013429

Problems in Physical Chemistry (Cont.)

SOV/4386

research in many facets of physical chemistry, including reaction kinetics, crystallography, spectroscopy, free energy studies, investigations of radiation effects in chemical reactions, low temperature studies, etc. Figures, tables, and references accompany the articles.

TABLE OF CONTENTS:

Pravednikov, A. N. The Influence of Polar and Steric Factors on Chain Transfer and Chain Growth Reactions During Polymerization

5

Temkin, M. I., N. M. Morozov, V. M. Pyzhev (Deceased), L. O. Apel'baum, L. I. Luk'yanova, and V. A. Demidkin. The Oxidation of Ammonia Over a Nonplatinum Catalyst

14

Pshezhetskiy, S. Ya., S. A. Kamenetskaya, Ye. I. Gribova, A. V. Pankratov, N. M. Morozov, I. N. Pospelova, A. Ya. Apin, V. N. Siryatskaya, N. A. Slavinskaya, and V. M. Cherednichenko. Kinetics of Decomposition, and the Explosion of Ozone

Card 2/5

Problems in Physical Chemistry (Cont.) SOV/4386	
Horiuchi, Juro (Japan). How to Find the Kinetic Equation of a Reversible Reaction	39
Kolotyrkin, Ya. M. The Effect of the Specific Adsorption of Anions on the Kinetics of Hydrogen Evolution and the Structuof the Metal-Solution Boundary	
Varshavskiy, Ya. M. The Nature and Mechanism of Electro- philic Hydrogen Exchange	61
Zvonkova, Z. V. Crystallochemical Data on the Nature of the Mutual Effect of Atoms	97
Romantsova, G. I. Investigation of the Effect of Inter- molecular Interaction on the Ultraviolet Absorption Spectra of Aromatic Compounds	107
Smagina., Ye. I., V. S. Kutsev and B. F. Ormont. Investigation of Equilibrium in the System Zirconium-Nitrogen at High Temperatures and the Dependence of the Free Energy of ZrNx Formation on Its Composition and Structure	118
Card 3/5	110

5 (4), 5 (3)

AUTHORS:

Yur'yev, V. M., Prayednikov, A. N., SOV,

SOV/20-125-6-36/61

Medvedev, S. S., Academician

TITLE:

The Influence of Oxidation Products on the Kinetics of the Oxidation of Cetane (Vliyaniye produktov okisleniya na

kinetiku okisleniya tsetana)

PERIODICAL:

Doklady Akademii nauk SSSR, 1959, Vol 125, Nr 6,

pp 1301-1302 (USSR)

ABSTRACT:

The oxidation of cetane takes place at 1400 in a closed system

with circulating oxygen. Figure 1 shows that, up to a

reaction yield of 25-30 %, the reaction develops

autocatalytically, after which it decreases rapidly and continues at a nearly constant rate above a reaction yield of

40-50 %. The concentration of peroxide compounds has a

maximum at a reaction yield of 25-30 %, after which it also decreases and becomes nearly constant at a reaction yield of 40-50 %. These phenomena are indicative of the fact that, in the course of oxidation, processes occur which reduce the rate of oxidation. As in the case of hydrocarbon oxidation, the system becomes divided into two layers in the course of

the process, an upper layer containing hydrocarbons and a

Card 1/2

The Influence of Oxidation Products on the Kinetics SOV/20-125-6-36/61 of the Oxidation of Cetane

lower one consisting of oxidation products, products of the lower layer were added to the cetane, which resulted in a reduction of the reaction rate (Fig 3). On the other hand, removal of the lower layer from the reaction vessel caused acceleration of the reaction. This proves that the reduction of reaction rate is caused by the accumulation of products which interrupt the development of the reaction. There are 3 figures.

ASSOCIATION:

Nauchno-issledovatel'skiy fiziko-khimicheskiy institut im. L. Ya. Karpova (Scientific Research Institute for

Physical Chemistry imeni L. Ya. Karpov)

SUBMITTED:

February 11, 1959

Card 2/2

TELESHOV, E.N.; PRAVEDNIKOV, A.N.; MEDVEDEV, S.S., akademik

Mechanism of polyisobutylene radiolysis. Dokl. AN SSSR 156 no.6: 1395-1398 Je '64. (MIRA 17:8)

1. Nauchno-issledovatel'skiy fiziko-khimicheskiy institut imeni L.Ya. Karpova.

TEIE SHOV, E.N.; TELESHOVA, A.S.; DESYATOVA, N.V.; PRAVEDNIKOV, A.N.; MEDVEDRV, S.S., akademik

Gas evolution and the formation of double bonds in the radiolysis of polyisobutylene. Dokl. AN SSSR 154 no.6:1402-1405 $^{\rm F}$ '64. (MIRA 17:2)

8/0020/64/154/006/1402/1405

AUTHOR: Teleshov, E. N.; Teleshova, A. S.; Desystova, N. V.; Pravedníkov, A. N.; Medvedev, S. S. (Academician)

TITLE: Gas release and formation of double bonds during radiolysis of polyisobutylene (PIB)

SOURCE: AN SSSR. Doklady*, v. 154, no. 6, 1964, 1402-1405

TOPIC TAGS: gas, double bond, radiolysis, polyisobutylene, Co sup 60, linear electron accelerator, free radical

ABSTRACT: Industrial polyisobutylene films, prepared by evaporation of weak solutions of a polymer in carbon tetrachloride were used. Co⁶⁰ (about 20 000 gm-equivalent Ra) and a linear electron accelerator (200 kev) were the source of ionizing radiation. Before irradiation, the films were evacuated to about 10⁻⁵ MM Hg during heating to 70C for 24 hours. The degree of destruction was estimated from viscosimetric data. The results indicate that the loss of free radicals in PIB at a temperature above the vitrification temperature is not accompanied by either a formation of gaseous products or the development of double bonds in the

Card 1/2

"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001342

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ACCESSION NR: AP401					
polymer. Orig. art.	has: 4 figs., 2	tables.			
ASSOCIATION: none					
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Peb AFFTC/ASD-3/SSD/RPL/ESD(t)/ESD(gs,/RAEM(c)/ESD(t)/RAEM(1)/SSD/BSD/AFWI/ASD(a)-5/
ACCESSION NR: AP4044277 AS(mp)-2 GG/RH/S/0192/64/005/C04/0627/0629

WW/JFW

AUTHORS: Teleshov, E.N.; Sharpaty*y, V.A.; Pravednikov, A.N.

Medvedev, S. S.

TITLE: Some changes in EPR spectra of irradiated polyisobutylene

SOURCE: Zhurnal strukturnoy khimii, v. 5, no. 4, 1964, 627-629

TOPIC TAGS: polyisobutylene, electron paramagnetic resonance electron, irradiation, uv. radiolysis, free radical, free radical recombination, polymer radiolysis

ABSTRACT: The irradiation of polyisobutylene (PIB) at liquid nitrogen temperature leads to accumulation of free radicals in it. The EPR spectrum of these radicals is a doublet with approximately 22 oersted splitting which is attributed to -C(CH₃)₂-CH-C(CH₃)₂ radical (I). In this work an attempt is made to obtain by the EPR method some additional information on the nature and properties of radical products which are formed during radiclysis of PIB. It was found that heating of PIB samples, irradiated with ~10²² eV/g dose of 1.6 mev electrons at -180 C leads, along with the destruction of primary radicals, to irreversible changes in EPR spectrum. In it the doublet is converted to a spectrum which consists of seven basic lines with addition of Card 1/3

fine structure. This spectrum may be ascribed to CH2-0. CH3 radical

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which is produced as a result of addition of isobutylene molecule to radical (I). Isobutylene is produced during radiolysis of PIB. During low temperature radiolysis of PIB radical (II) is not detectable by the EPR method because radicals which are formed during breakage of the radical chain immediately enter recombination and disproportionation reactions. Irradiation of PIB at -60C (above vitrification temperature of the polymer) with simultaneous registration of EPR spectra enables one to find radicals (I) as well as radicals (II). PIB irradiated with UV at -60C for 5 min produces EPR spectrum similar to that of a mixture of PIB and isobutylene irradiated with electrons. It is suggested that under the influence of UV, isomerisation of primery radicals may take place:

Orig. art. has: 3 figures

Card 2/3

"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001342

L 15706-65 ACCESSION NR: AP4044277 ASSOCIATION: Fiziko-khimicheskiy institut im. L. Ya. Karpova (Institute of Physical Chemistry)						
SUBMITTED: 28Dec63		ENCL: 00				
SUB CODE: OPNP	NR RBF 80V: 004	OTHER: 006				
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Card 3/3						

\$/0020/64/156/003/0626/0629

AUTHORS: Sharpaty*y, V.A.; Aptekar', Ye.L.; Zakatova, N.V.; Pravedni-kov, A.N.

TITLE: Radiolysis of polyamides

SOURCE: AN SSSR. Doklady*, v. 156, no. 3, 1964, 626-629

TOPIC TAGS: polyamide, radiolysis, mechanism, kinetics, radical radiolysis product, EPR method, radical mechanism, molecular cleavage, carbon hydrogen bond rupture, butyricbutyroamide, chromophoric group

ABSTRACT: This study was conducted to obtain information about the initial stages of the radiolysis of the polyamides -CONH(CH₂)_n CONH(CH₂)_n or -CONH(CH₂)_nNHCO(CH₂)_mCONH (where n and m can be 4 to 10) and their low molecular analogs. CO and H₂ are formed on radiolysis of polyamides, with the formation of H₂ being independent of radiolysis temperature and proportional to the dosage. The nature and kinetics of the accumulation of radical radiolysis products were studied by the EPR method. The yield of accumulated radicals is almost independent of the type of sample (resin or fiber) or of radiolysis temperature, and increases with the number of methyl groups in Card | 1/3

the polymer chain. The radical -CONHĆ H₂ is presumed to be formed by rupture of the C-H bond in the methylene groups. The atomic hydrogen reacts with the polymeric material pulling away a hydrogen atom from the \$\pi\$-methylene bonds. On illumination with visible light for 15-20 minutes the EPR spectrum changes sharply, the sample coloring intensity is increased and no gas is evolved. Further illumination has no effect. Apparently the radical formed also exists as CH₂ CONHCH=CHCH₂ with the number of the chromophoric groups being retained but rearranged. Mass spectrometric analysis of the radiolysis products of butyroamide of butyric acid led to the assumption of the following radiolysis scheme:

 $C_3H_3 - C = O - NH - CH_4 - C_3H_3 - C_3H_3CO - NH - CH - C_3H_3 - C_3H_3 - CH_3 - CH_3 - CH_3H_3$, $C_3H_3CO - C_3H_3 + CO. 2C_3H_3 = CH_3CH_3CH_3 + CH_3CH = CH_3$.

Since in the radiolysis of the polyamides and of the low molecular analog the amount of $\rm H_2$ exceeds that of CO, and the amount of crosslinkage does not cover the difference between the two, it was concluded that $\rm H_2$ is formed during radiolysis by the radical mechanism and by molecular cleavage from two adjacent carbon atoms or from the

Curd 2/3

ACCESSION NR: AP4038527

nitrogen and carbon atoms near the carbonyl group. Thus the processes of H and of CO formation during the radiolysis of polyamides are independent to some degree. "The authors thank M.K. Dobrokhotov, A.V. Sharov, D.M. Margolin, B.V. Maslova and K.G. Yanov for help in the work." Orig. art. has: 1 table, 4 figures and 1 equation.

ASSOCIATION: Fiziko-khimicheskiy institut im. L. Ya Karpova (Physical

Chemical Institute)

SUBMITTED: 18Dec63 ENCL: 00

SUB CODE: NP, OC NR REF SOV: 001 OTHER: 005

Cord 3/3

KARDASH, I.Ye.; PRAVEDNIKOV, A.N.; MEDVEDEV, S.S., akademik

Thermal degradation of polyethylene terephthalate. Bokl.
AN SSSR 156 no. 3:658-661 '64. (MIRA 17:5)

1. Fiziko-khimicheskiy institut im. L.Ya.Karpova.

·s/0020/64/156/006/1395/1398

AUTHOR: Teleshov, E. N.; Pravednikov, A. N.

ACCESSION NR: AP4041404

TITLE: The mechanism of polyisobutylene radiolysis

SOURCE: AN SSSR. Doklady*, v. 156, no. 6, 1964, 1395-1398

TOPIC TACS: polyisobutylene, radiolysis, radiolysis mechanism, isobutylene, free radical formation, EPR spectra, polymer chain rupture, vinyl type double bond, disproportionation reaction, recombination reaction, cross linkage, depolymerization

ABSTRACT: A mechanism for the reactions occurring during the radiolysis of polyisobutylene (PIB) was proposed based on the examination of the free radicals formed by irradiation FIB with cobalt-60 or accelerated electrons (1.6 MeV) at different temperatures under about 10-5 mm Hg. The EPR spectra show that at temperatures below about 65C the free radical formation is proportional to the irradiation dosage; at higher temperatures the radical yield is lowered. At liquid nitrogen temperature the radical C(CH₃)₂-CH-C(CH₃)₁ (I) is formed, but examination of the PIB radiolysis end products indicates other structures are formed which react at lower temperatures.

These result from the rupture of the polymer chain:

or breaking off of hydrogen or a methyl group to form:

The formation of the vinyl type double bonds is explained by the disproportionation reaction of II and III at liquid nitrogen temperature:

Isobutylene is formed by the radiolysis of PIB, the depolymerization increasing only above room temperature. PIB does not crosslink, indicating radicals I, IV and V do not enter a recombination reaction. It is concluded that the destruction of PIB in the early stages of radiolysis proceeds by reaction (1) with subsequent disproportionation (reaction 2). Since reaction (1) is independent of temperature, the temperature dependence of PIB destruction is explained by a change in the relative rate of disproportionation and recombination of radicals II and III at different temperatures. Orig. art. has: 3 figures and 3 formulae.

ASSOCIATION: Nauchno-issledovatel'skiy fiziko-khimicheskiy institut im. L. Ya. Karpova (Scientific Research Institute of Physical Chemistry)

SUBMITTED: 26Mar64

ENCL: 00

SUB CODE: GC

NR REF SOV: 004

OTHER: 007

Card 3/3

YUR'YEV, V.M.; TELESHOVA, A.S.; APTEKAR', Ye.L.; ARDASHNIKOV, A.Ya.; REZNIKOVA, O.Ya.; PRAVEDNIKOV, A.N.

Use of ion-sorption ESh-1 pumps in the MI-1305 mass-spectrometer. Zav.lab. 30 no.3:375-376 '64. (MIRA 17:4)

1. Nauchno-issledovatel'skiy fiziko-khimicheskiy institut imeni Karpova.